

Network analysis of Kolmogorov's condition

For a given transition rate matrix R ,
we can determine whether K's cond'n
holds, by examining fundamental cycles.

These offer a kind of linear decomposition
of a connected graph G . In a stationary
state, once we specify the stationary
current associated w/ each fundamental
cycle, we can determine all the stationary
currents J_{ij}^s .

see Schnakenberg, Rev Mod Phys, 1976
for more details.

$N = \#$ of nodes in G

$E = \#$ of edges

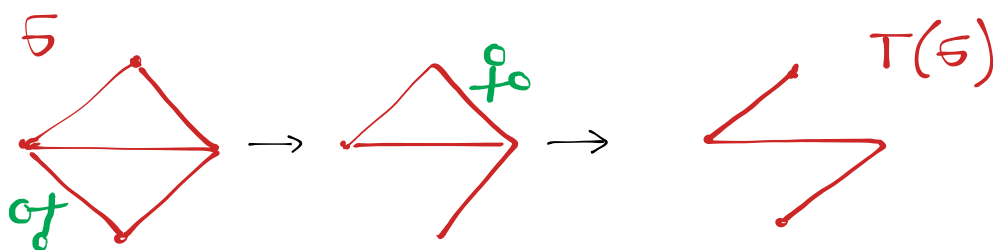
(= $\#$ of non-zero R_{ij} 's above diagonal)

then $C = E - N + 1 = \#$ of fund. cycles

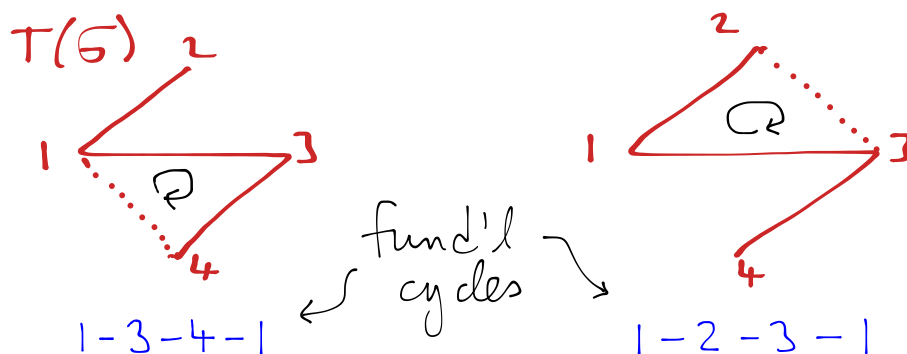
If $C=0$, then $G = \text{tree}$

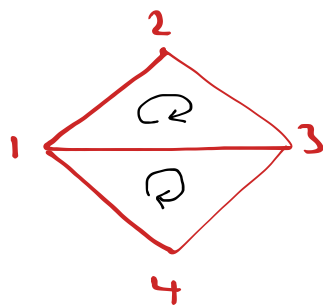
Let's assume $C > 0$. Then there are 2 steps for constructing a set of fund'l cycles.

- ① Construct a maximal tree $T(G)$ by removing C edges from G , without breaking connectivity.

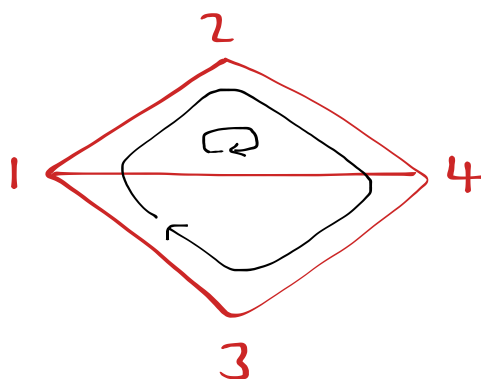
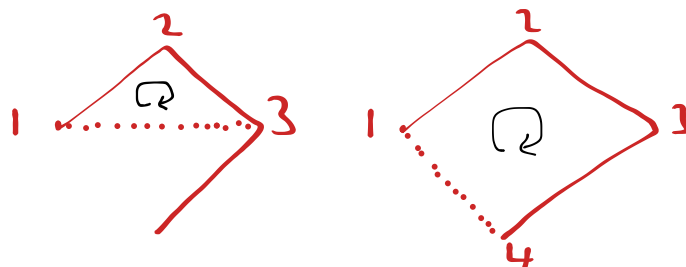
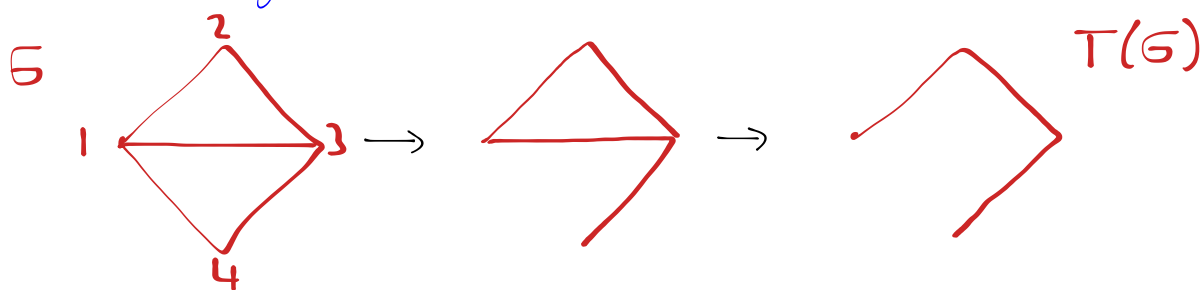


- ② For each edge removed, we obtain a fund'l cycle by replacing that edge only:





This construction is not unique: if we had chosen to remove edges 1-4 & 1-3 (instead of 1-4 & 2-3) then:



...but this G always has 2 fund. cycles.

To determine whether Kolmogorov's condition is satisfied, we first decompose G into $C = E - N + 1$ fund'l cycles.

Then for each fund. cycle $C = i_1 - i_2 - \dots - i_M - i_{M+1} = i_1$, we determine the affinity

$$A_c = \sum_{k=1}^M \ln \frac{R_{i_{k+1} i_k}}{R_{i_k i_{k+1}}} \quad (k_B T = 1)$$

Kolmogorov's condition (& therefore det. bal.) is satisfied if and only if $A_c = 0$ for every fundamental cycle.

In other words: no need to check K's condition for every simple cycle in G , but only for a set of fundamental cycles.

If R satisfies detailed balance, then a system evolving under $\frac{d\vec{p}}{dt} = R\vec{p}$ relaxes to a stationary state $\vec{\pi}$ with no currents:

$J_{ij}^s = 0 \quad \forall i, j$. If we are modeling a system in contact with a heat bath, then this stationary state is an equil state:

$$\pi_i = \frac{1}{Z} e^{-\beta E_i}, \quad \beta^{-1} = k_B T$$

There are two ways of bringing nonequilibrium behavior into the picture:

① Consider a rate matrix R that violates det. bal., & study the non-equilibrium stationary state $\vec{\pi}$.

*② Consider time varying transition rates:

$$R = R(\lambda), \quad \lambda = \lambda(t)$$

but assume $R(\lambda)$ satisfies det. bal. for all λ .

Assume $R(\lambda)$ satisfies det. bal. $\forall \lambda$,
 & we are using R to model a system
 in contact w/ a thermal reservoir.

Then
$$\ln \frac{R_{ij}}{R_{ji}} = \frac{E_j - E_i}{k_B T} = \ln \frac{\pi_i}{\pi_j}$$

i.e. we can use $R(\lambda)$ to assign energies
 $E_1(\lambda), \dots, E_N(\lambda)$ to the discrete
 states of the system. (Assume
 the zero of energy is set by
 the physical context.)

Now imagine a process during which
 λ is varied with time, $0 \leq t \leq \tau$.

1 realization:

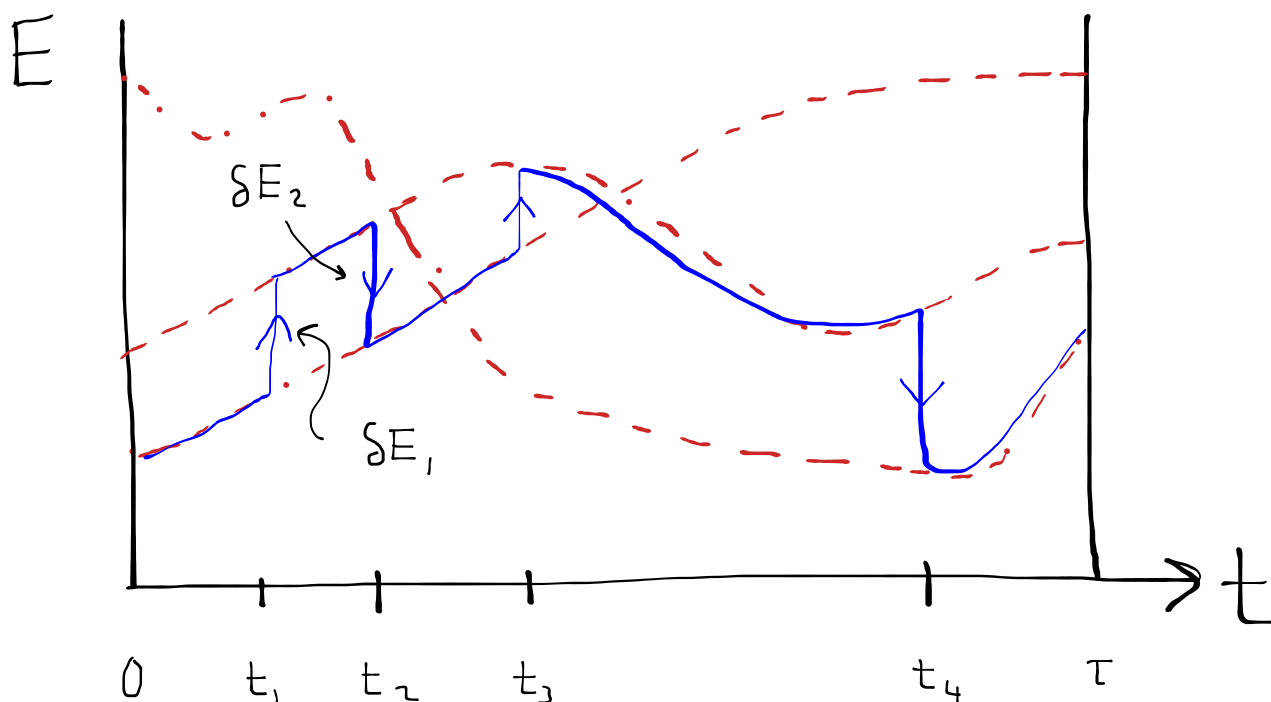
$$(t=0) \quad i_0 \xrightarrow{t_1} i_1 \xrightarrow{t_2} i_2 \cdots \xrightarrow{t_K} i_K \quad (t=\tau)$$

t_i = time of i^{th} transition.

K differs from 1 realization to another.

Time-dependent energy: $E(t) = E_{i(t)}(\lambda(t))$

$\uparrow$ microstate
@ time t



dashed lines = $E_n(\lambda(t))$'s

solid line depicts trajectory $i(t)$
(one realization)

$E(t)$ changes w/time due to both:

- transitions from state to state
- the time-dependence of λ

$$\begin{aligned}
\Delta E &= E_{\text{final}} - E_{\text{initial}} = E(i_K, \lambda(\tau)) - E(i_0, \lambda(0)) \\
&= \underbrace{\sum_{k=1}^K \delta E_k}_{\substack{\text{interpret} \\ \text{as } \underline{\text{heat}}}} + \underbrace{\int_0^\tau dt \dot{\lambda}(t) \frac{\partial E}{\partial \lambda}(i(t), \lambda(t))}_{\substack{\text{interpret} \\ \text{as } \underline{\text{work}}}} \\
&\quad \downarrow \qquad \qquad \downarrow \\
&= Q + W \quad (1^{\text{st}} \text{ Law})
\end{aligned}$$

Ensemble: $\langle E \rangle = \sum_i p_i(t) E_i(\lambda(t))$

$$\begin{aligned}
\frac{d}{dt} \langle E \rangle &= \sum_i \left(\dot{p}_i E_i + \dot{\lambda} p_i \frac{\partial E_i}{\partial \lambda} \right) \\
&= \left(\sum_{ij} R_{ij} p_j E_i \right) + \left(\dot{\lambda} \sum_i p_i \frac{\partial E_i}{\partial \lambda} \right) \\
&= \langle \dot{Q} \rangle + \langle \dot{W} \rangle \\
&\quad \frac{d}{dt} \langle Q \rangle_t \qquad \qquad \frac{d}{dt} \langle W \rangle_t
\end{aligned}$$

To get to the 2nd Law, let's first consider the case of fixed λ .

$\mathcal{D}(\vec{p} | \vec{\pi})$ = relative entropy of current state $\vec{p}$
w/ respect to equil. state $\vec{\pi}$.

$$= \sum_{i=1}^N p_i \ln \frac{p_i}{\pi_i} \geq 0$$

$$\frac{d}{dt} \mathcal{D} = \sum_i \left(\dot{p}_i \ln \frac{p_i}{\pi_i} + p_i \cdot \frac{1}{p_i} \dot{p}_i \right)$$

$$= \sum_{ij} R_{ij} p_j \ln \frac{p_i}{\pi_i}$$

$$= \sum_{i \neq j} (R_{ij} p_j - R_{ji} p_i) \ln \frac{p_i}{\pi_i}$$

$$= \frac{1}{2} \sum_{i \neq j} (R_{ij} p_j - R_{ji} p_i) \ln \frac{p_i}{\pi_i} + \frac{1}{2} \sum_{i \neq j} (R_{ji} p_i - R_{ij} p_j) \ln \frac{p_j}{\pi_j}$$

$$\uparrow \quad i \leftrightarrow j \quad \uparrow$$

$$= \frac{1}{2} \sum_{i \neq j} (R_{ij} p_j - R_{ji} p_i) \ln \left(\frac{p_i \pi_j}{\pi_i p_j} \right)$$

$$= \sum_{i < j} (R_{ij} p_j - R_{ji} p_i) \ln \left(\frac{R_{ji} p_i}{R_{ij} p_j} \right)$$

use
det. bal.

$$= - \sum_{ij} \underbrace{(a_{ij} - a_{ji})}_{+ \text{ or } -} \underbrace{\ln \left(\frac{a_{ij}}{a_{ji}} \right)}_{+ \text{ or } -}$$

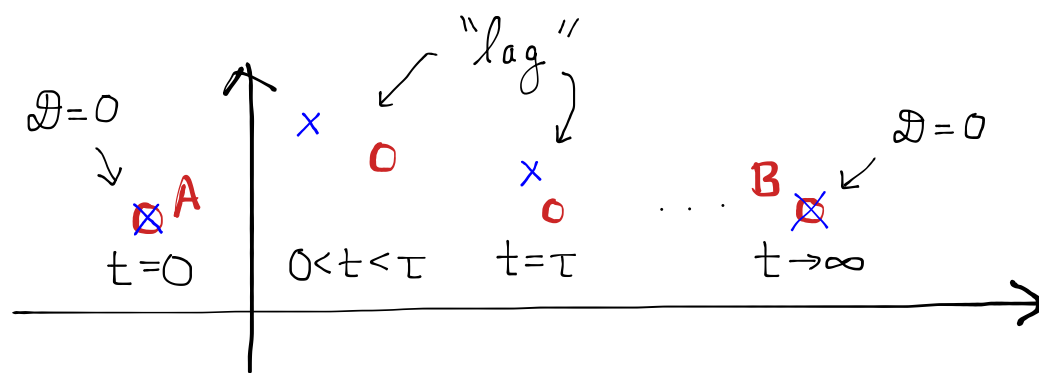
$$a_{ij} = R_{ij} p_j$$

$$\leq 0$$

As with diffusive processes, $\mathcal{D}(\vec{p}|\vec{\pi})$ provides a measure of "distance" from equilibrium. This distance decreases w/time, when λ is held fixed.

Now consider the process:

1. Start in equil., $\lambda = A$: $\vec{p}(0) = \vec{\pi}_A$
2. $\lambda: A \rightarrow B$ as $t: 0 \rightarrow \tau$
3. Hold $\lambda = B$ fixed for $t > \tau$.
 $\vec{p}(t) \rightarrow \vec{\pi}_B$ as $t \rightarrow \infty$.



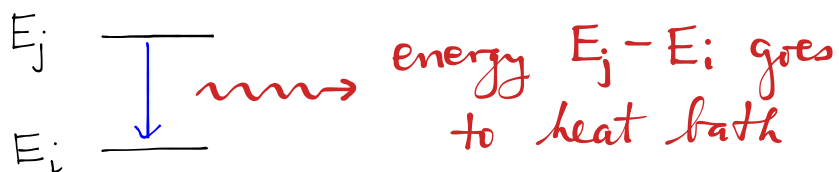
$$\begin{cases} \times & = \text{state of system, } \vec{p}(t) \\ \circ & = \text{equilibrium state, } \vec{\pi}_{\lambda(t)} \end{cases}$$

Now let's make a connection to entropy.

Start by assuming fixed λ , & take $\vec{p}(0) \neq \vec{\pi}$.

$$\begin{aligned}
 0 &\leq -\frac{d}{dt} \mathcal{D}(\vec{p} | \vec{\pi}) \\
 &= \sum_{i < j} (\underbrace{R_{ij} p_j - R_{ji} p_i}_{\text{red bracket}}) \ln \frac{R_{ij} p_i}{R_{ji} p_j} \\
 &= \sum_{i < j} J_{ij} \ln \frac{R_{ij}}{R_{ji}} + \sum_{i < j} J_{ij} \ln \frac{p_i}{p_j} \quad \text{show!} \\
 &= \beta \sum_{i < j} J_{ij} (E_j - E_i) + \frac{d}{dt} \left[\underbrace{-\sum_i p_i \ln p_i}_{S(\vec{p}), k_B=1} \right]
 \end{aligned}$$

J_{ij} = avg. rate of transitions $j \rightarrow i$



$$\therefore \beta \sum_{i < j} J_{ij} (E_j - E_i) = \frac{1}{T} \times \text{avg. rate of energy flow to reservoir}$$

$$= -\frac{\langle \dot{Q} \rangle}{T} = \frac{dS_{\text{res}}}{dt}$$

Combining terms :

$$0 \leq \frac{dS_{\text{res}}}{dt} + \frac{dS}{dt} \quad 2^{\text{nd}} \text{ Law!}$$

This derivation can be extended to processes during which λ varies with time.

We still get

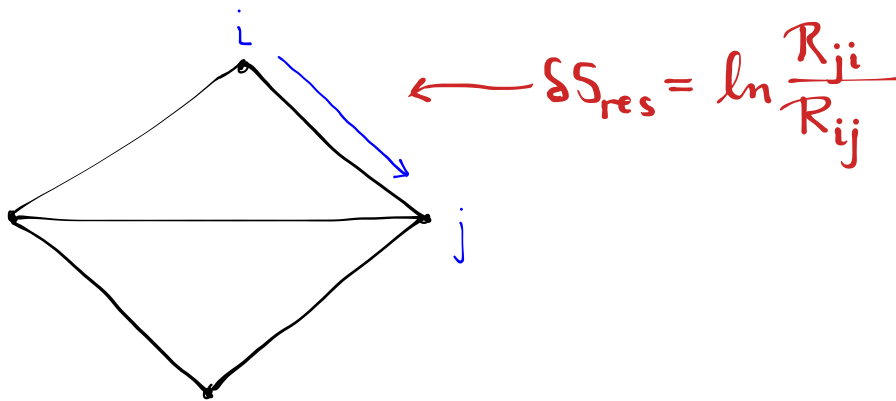
$$0 \leq -\frac{\langle \dot{Q} \rangle}{T} - \frac{d}{dt} \sum_i p_i \ln p_i = \frac{dS_{\text{res}}}{dt} + \frac{dS}{dt}$$

At the single trajectory level :

$$(t=0) \quad i_0 \xrightarrow{t_1} i_1 \xrightarrow{t_2} \dots \xrightarrow{t_K} i_K \quad (t=\tau)$$

$$\Delta S_{\text{res}} = \frac{\text{energy absorbed by reservoir}}{T}$$

$$= \beta \sum_{l=1}^K [E(i_{l-1}) - E(i_l)] = \sum_{l=1}^K \ln \left(\frac{R_{i_l i_{l-1}}}{R_{i_{l-1} i_l}} \right)$$



When the system makes a transition $i \rightarrow j$, the change in the entropy of the reservoir is $\ln \frac{R_{ji}}{R_{ij}}$.

We have used detailed balance to derive

$$\Delta S_r = \sum_{l=1}^K \ln \left(\frac{R_{i_l i_{l-1}}}{R_{i_{l-1} i_l}} \right), \quad \delta S_r(i \rightarrow j) = \ln \frac{R_{ji}}{R_{ij}}$$

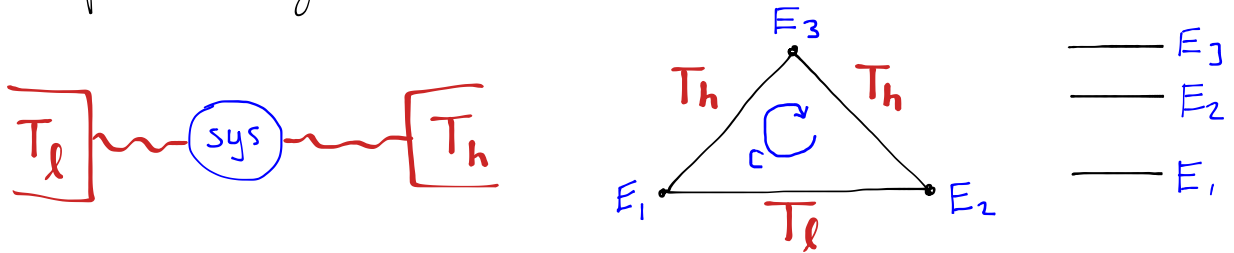
but the result remains valid even if R violates detailed balance. *

$\Delta S_r, \delta S_r$ = net change in entropy
off all thermal surroundings
(might include multiple reservoirs)

Instead of trying to justify this statement from microscopic principles, let's see how it works in two illustrative examples.

* This is a central principle of
Stochastic Thermodynamics
(Seifert, Rep. Prog. Phys. 2012)

Example 1 system + 2 heat baths



$$\frac{R_{31}}{R_{13}} = e^{-\beta_h(E_3 - E_1)}, \quad \frac{R_{23}}{R_{32}} = e^{-\beta_h(E_2 - E_3)}, \quad \frac{R_{12}}{R_{21}} = e^{-\beta_l(E_1 - E_2)}$$

$$\begin{aligned} A_c &= \ln \left(\frac{R_{12}}{R_{21}} \cdot \frac{R_{23}}{R_{32}} \cdot \frac{R_{31}}{R_{13}} \right) \\ &= -\beta_l(E_1 - E_2) - \beta_h(E_2 - E_3) - \beta_h(E_3 - E_1) \\ &= (E_2 - E_1) \cdot \left(\frac{1}{T_l} - \frac{1}{T_h} \right) > 0 \end{aligned}$$

$$A_c = \Delta S_r(1 \rightarrow 3 \rightarrow 2 \rightarrow 1) > 0$$

... every time the system proceeds
 thru the cycle c , the net result
 is a transfer of energy $E_2 - E_1$
 from the hot bath to the cold bath.

(In the stationary state, $J_{31} = J_{23} = J_{12} > 0$.)

The temperature difference $T_h - T_l$ is
 like a force that drives a CW current.)